A Marked-up copy of the Claims

- 1. (amended) A process for the fractionation or purification of <u>polydisperse</u> water soluble polymers <u>that are composed of repeating units that do not contain functional groups capable of carrying a charge at neutral pH comprising:</u>
 - a. dissolving a known amount of the polymer in water to form a clear solution,
 - b. equilibrating the solution at a temperature that allows the formation of an opaque solution when an aqueous soluble, <u>multivalent</u> salt is added to the solution,
 - c. adding an aqueous soluble multivalent [extraction]salt to the solution,
 - d. maintaining the solution at the temperature of step b until two phases form in the solution, such that the lower phase contains the lower molecular weight polymer molecules,
 - e. removing the lower phase,
 - f. replacing the volume of the lower phase by adding water to the remaining solution,
 - g. repeating steps b through f the required number of times to cause the desired fractionation.
 - h. isolating the upper phase of the final extraction,
 - i. removing the water and <u>multivalent</u> [extraction] salt to yield a fractionated polymer having a higher average molecular weight than the polydisperse polymer.
- 2. (amended) The process of claim 1 wherein the concentration of the <u>multivalent aqueous</u> soluble extraction salt is sufficient to cause two distinct phases to form at the selected temperature.
- 3. (amended) The process of claim 2 wherein the <u>multivalent</u>, aqueous soluble extraction salt is selected from the group consisting of sulfate, citrate, [or] <u>and</u> phosphate salts.
- 4. (amended) The process of claim 2 wherein the <u>polydisperse</u> water soluble polymer is [a polyol or] a polyether.
- 5. (withdrawn) The process of claim 4 wherein the polyol polymer is composed of ethylene oxide monomers joined by ether linkages.
- 6. (amended) The process of claim 4 wherein the [polyol] <u>polyether</u> polymer is a polyoxyalkylene block copolymer.
- 7. The process of claim 6 wherein the polyoxyalkylene block copolymer is a poloxamer.

- 8. (withdrawn) The process of claim 6 wherein the polyoxyalkylene block copolymer is a poloxamine.
- 9. The process of claim 4 wherein the <u>multivalent</u>, aqueous <u>soluble</u> extraction salt is ammonium sulfate.
- 10. (amended) The process of claim 9 wherein the concentrations of the <u>polydisperse [polyol]</u> polymer and ammonium sulfate, and the extraction temperature are adjusted so that the lower molecular weight [polyol] polymer molecules partition into the high salt concentration (lower) phase and the higher molecular weight [polyol] polymer molecules partition into the low salt concentration (upper) phase of the aqueous fractionation medium.
- 11. The process of claim 10 wherein the concentration of ammonium sulfate is about 5 to about 25% by weight of the solution.
- 12. (amended) The process of claim 10 wherein the concentration of the [polyol] <u>polydisperse</u> polymer may be up to 10% by weight of the extraction solution.
- 13. (amended)The process of claim 10 wherein the extraction temperature ranges from about –5 to about 30 [degree] degrees C.
- 14. (amended)The process of claim 10 wherein the extraction temperature ranges from about -2 to about 10 [degree] degrees C.
- 15. (amended) The process of claim 1 herein the polydispersity of the [resulting polyol] fractionated water soluble polymer is reduced.
- 16. (amended) The process of claim 1 wherein the viscosity of aqueous solutions of the [resulting polyol] <u>fractionated water</u> soluble polymer is increased between about 25 and about 40 [degree] <u>degrees</u> C.
- 17. The process of claim 1 wherein the viscosity of aqueous solutions of the [resulting polyol] fractionated water soluble polymer increases rapidly over a narrow temperature range.
- 18. (amended) [The process of claim 1 wherein the aqueous soluble extraction salt is replaced by an aqueous soluble polymer that is incompatible with the polymer that is to be fractionated.] A process for the fractionation or purification of polydisperse water soluble polymers that are composed of repeating units that do not contain functional groups capable of carrying a charge at neutral pH comprising:

- a. dissolving a known amount of said polymer in water to form a clear solution,
- b. equilibrating said solution at a temperature that allows the formation of an opaque solution when a second aqueous soluble polymer that is incompatible with the polymer to be fractionated is added to the solution,
- c. <u>adding to the solution a second aqueous soluble polymer that is incompatible with the polymer to be fractionated.</u>
- d. maintaining the solution at the temperature of step b until two phases form in the solution.
- e. removing the phase that contains the incompatible polymer and the lower molecular weight molecules of the polydisperse water soluble polymer,
- f. replacing the volume removed by adding water to the remaining solution,
- g. repeating steps b through f the required number of times to cause the desired fractionation,
- h. isolating the phase of the final extraction that contains the higher molecular weight molecules of the water soluble polymer to be fractionated.
- i. removing the water to yield a fractionated polymer having a higher average molecular weight than the polydisperse polymer.
- 19. (amended) The process of claim 18 wherein the concentrations of the <u>polydisperse water</u> <u>soluble</u> polymer to be fractionated, and the incompatible polymer, and the extraction temperature are adjusted so that the extraction system forms two phases, and the lower <u>molecular weight polymer molecules of the polydisperse water soluble polymer partition into the phase that contains the incompatible polymer.</u>
- 20. (withdrawn) A purified water soluble polymer made by the process of claim 1.
- 21. (new) The process of claim 2 wherein the polydisperse, water soluble polymer is a polyol.